

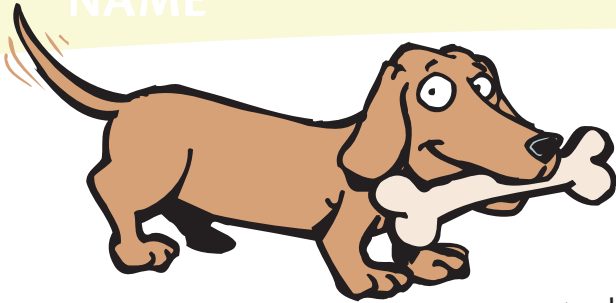
Answers for Kids

BIBLE CURRICULUM

How about a Date? Part 1

LESSON 17

NAME _____



“By the word of the LORD the heavens were made, and all the host of them by the breath of His mouth” (Psalm 33:6).

How many times have you heard something like, “This animal lived 50,000 years ago” or “This person died 20,000 years ago”? Have you ever wondered how the scientists come up with the age of the bone? After all, the scientists haven’t been around that long, have they?

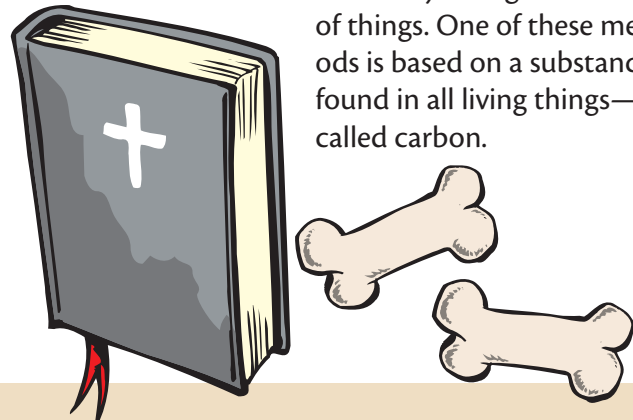
There are a number of different ways to figure out how old an object is.

Of course, the best method is

to check the account of a reliable eyewitness, if one is available. The Bible is such a record. Since it is the Word of God, we can trust it to tell us the truth about the past. Carefully studying what the Bible says, we find that the universe has an age of around 6,000 years, and that

a world-changing, global Flood occurred about 4,300 years ago.

Those who don’t accept the biblical account of history look for other ways to figure out the age of things. One of these methods is based on a substance found in all living things—it’s called carbon.



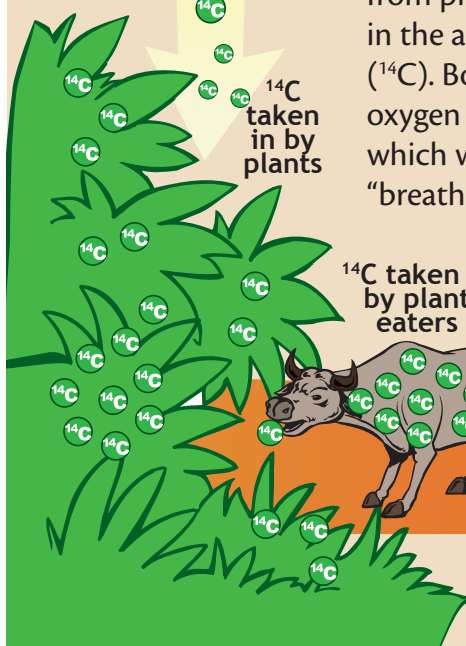
How Carbon Dating Works



There are two basic forms of carbon: one that occurs naturally, called carbon-12 (^{12}C), and one that forms from processes acting on nitrogen in the atmosphere, called carbon-14 (^{14}C). Both of these combine with oxygen to form carbon dioxide (CO_2), which we breathe out and plants “breathe” in. When a cow eats grass, its

body absorbs the carbon (both ^{12}C and ^{14}C) in the plant.

When the cow dies, it stops taking in carbon (for obvious reasons). The amount of ^{12}C in the cow’s body stays the same after death, but the amount of ^{14}C changes because it returns to nitrogen.



^{14}C
taken
in by
plants

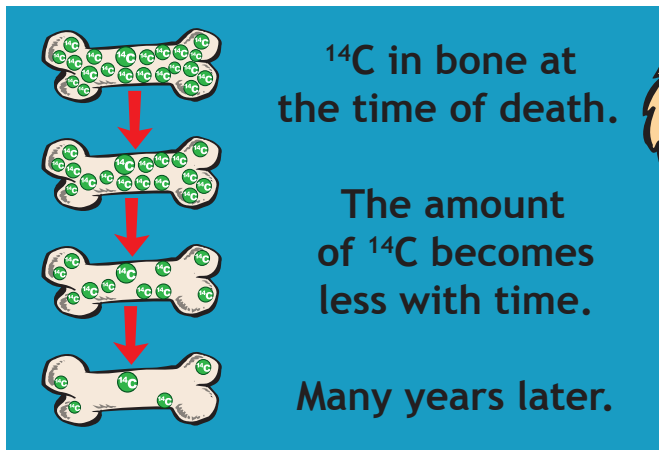
^{14}C taken in
by plant
eaters

^{14}C taken in by
eaters of the
plant eaters

**At
death,
intake
stops.**



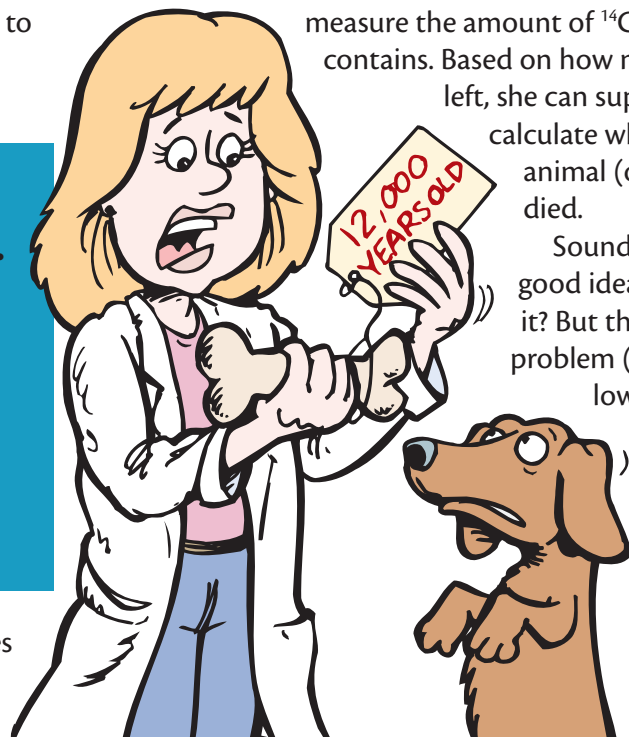
As time goes on, the amount of ^{14}C continues to decrease until nothing is left, which is supposedly about 50,000 years later.



When a paleontologist (a scientist who studies bones) finds a bone (or a piece of wood), she can

measure the amount of ^{14}C and ^{12}C it contains. Based on how much ^{14}C is left, she can supposedly calculate when the animal (or plant) died.

Sounds like a good idea, doesn't it? But there's a problem (see below).



The problem is

There are many things that affect how much ^{14}C an animal (or a person or plant) has in it when it dies. This changes how long ago the animal appears to have died.

For instance, plants don't take in as much ^{14}C as scientists expect. So, after they die, there is less ^{14}C in the plants to change back to nitrogen. This makes the plant appear to have died many more years ago than it actually did (for example, the plant might appear to be, say, 3,000 years old, rather than 2,000).

Also, the amounts of ^{14}C and ^{12}C in the atmosphere haven't been constant throughout history (for instance, Noah's Flood lowered the total amount of available carbon by burying lots of animals and plants, while the atmosphere continued to produce ^{14}C). So something that lived (and died) when the proportion of ^{14}C was

less than normal would appear to have died more years ago than it actually did (for example, it might give an age of 3,000 years before the present, rather than its true age of 2,000 years).

Even many archaeologists don't think "carbon dating" is completely accurate all the time.

When these (and other) problems are taken into account, a scientist can interpret the result of the carbon dating within a biblical time frame, but even so, these results can not be used to *prove* the age of once-living things.

